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Response surface modeling and optimization of process parameters for aqueous extraction of pigments from prickly pear (*Opuntia ficus-indica*) fruit

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ARTICLE INFO

Article history:
Received 14 May 2012
Received in revised form
13 June 2012
Accepted 15 June 2012
Available online 26 June 2012

Keywords:
Pigment
Prickly pear
Box—Behnken design
Extraction
Optimization
Response surface methodology

ABSTRACT

The objectives of this study are to determine the optimum conditions for the aqueous extraction of pigments from prickly pear fruit using a three-level three-factor Box—Behnken design under response surface methodology. Pigments were extracted from prickly pears using water as solvent at different extraction temperature (30–50 °C), time (20–120 min) and mass of fruit (0.5–1.5 g). The experimental data obtained were analyzed by Pareto analysis of variance (ANOVA) and fitted to a second-order polynomial equation using multiple regression analysis. An optimization study using Derringer's desired function methodology was performed and the optimal conditions based on both individual and combinations of all responses (extraction temperature of 40 °C, time of 115 min and mass of 1.44 g) were determined. At this optimum condition, the total betacyanin and betaxanthin content were found to be 13.4354 mg/100 g and 24.2922 mg/100 g with desirability value of 0.917. The experimental values agreed with those predicted values.

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1. Introduction

Opuntia spp., the largest genus of the Cactaceae family, includes manifold species and varieties and most known cactus fruits were used for various kinds of food products [1]. Different parts of the Opuntia ficus-indica are utilized as fruits and vegetable for human consumption, fodder for cattle, and raw materials for various industries to prepare plywood, soap, dyes, adhesives and glue, pharmaceutical products for treating diabetes and various other disorders, and cosmetics such as shampoo, cream, and body lotions, etc. [2]. In addition to the excellent quality and flavor of the fresh fruit, the young leaves serve both as a vegetable and salad dish and the immature fruit is used to make mock gherkins.

Cactus pear fruit (prickly pear) is a fleshy berry, varying in shape, size, and color and has a consistent number of hard seeds. It is characterized by a high sugar content (12–17%) and low acidity (0.03–0.12%). It has higher vitamin C, potassium, calcium and phosphorous and low sodium [3]. The prickly pears are considered as a rich source of yellow—orange betaxanthins and red—violet betacyanins and the red and purple colored prickly pears contain high amounts of total phenols and purple skinned fruits contain the

highest amounts of flavonoids which are responsible for the color of *Opuntia* spp., and having radical-scavenging and reducing properties [4–7]. The betalain content of the cactus pear fruit is found to be having an application in the low acid foods as natural colorants [8]. Cactus pear fruit pulp exhibits a high pH value (5.6–6.5) and total soluble solids content ranging from 11 to 17 °brix and this property making the pulp highly susceptible to microbial spoilage [9–11].

The extraction of bioactive compounds from permeable solid plant materials using solvents constitutes an important step in the manufacture of phytochemical-rich products. The application of this low-cost technology to obtain molecules to be used as food additives or nutraceutical products is an appropriate strategy for the exploitation of some non-timber products found in rain forests such as the fruits of the palm [12]. From the previous findings, many factors such as solvent composition, extraction time, extraction temperature [13], solvent to solid ratio [14] and extraction pressure [15], among others, may significantly influence the extraction efficacy of diverse natural products [16–18].

Response surface methodology (RSM), a powerful mathematical and statistical technique, has been effectively used in testing multiple process factors and their interactive effects [19,20]. RSM can help in investigating the interactive effect of process variables and in building a mathematical model that accurately describes the overall process [21]. Box—Behnken design is a spherical, revolving

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response surface methodology (RSM) design that consists of three interlocking 2^2 factorial design having points, all lying on the surface of a sphere surrounding the center of the design [22]. Several studies have been used successfully the RSM to model and optimize the biochemical and biotechnological processes related to food systems [23–27].

The extraction and purification of phytochemicals from natural sources are needed since these bioactives are often used in the preparation of dietary supplements, nutraceuticals, functional food ingredients and additives, pharmaceutical and cosmetic products [28]. In this study, the effects of various process parameters such as extraction temperature, time and mass of fruit on aqueous extraction of pigments (betacyanin and betaxanthin) from prickly pear fruit were investigated by applying Box—Behnken design under response surface methodology and the optimal conditions were determined to achieve the maximal pigments extraction.

2. Materials and methods

2.1. Fruit

Common name: Indian fig, Tuna and Nopal English name: Barbary fig and Prickly pear

Botanical name: Opuntia ficus-indica

Kingdom: Plantae Order: Caryophyllales Family: Cactaceae Used part: Fruits

The fruit of prickly pears, commonly called cactus fruit, cactus fig and Indian fig or tuna in Spanish, is edible, although it has to be peeled carefully to remove the small spines on the outer skin before consumption. The prickly pear contains betalain, betanin and indicaxanthin, with highest levels in their fruits [29]. The fruit of prickly pears is often used to make candies, jelly, or drinks such as vodka or lemonade and its pulp and juice have been used to treat numerous maladies, such as wounds and inflammations of the digestive and urinary tracts in folk medicine [30]. The aqueous extracts of cactus pear fruits exhibited a marked antioxidant capacity in several in vitro assays, including the oxidation of red blood cell membrane lipids and the oxidation of human LDLs induced by copper and 2,2'-azobis (2-amidinopropane-hydro chloride) [31].

2.2. Fruit preparation and aqueous extraction of pigments

Freshly harvested prickly pear fruits (*Opuntia ficus-indica* – with similar maturity and weight) were used as raw materials and they were procured from the local venders near Erode, TamilNadu, and were stored at 4 °C prior to the experiments. The thick layer or skin and thorns were peeled from the fruits manually and the fruits were washed thoroughly in running tap water to get rid of any impurities adhered to the surface of the fruit and cut into required weight for the extraction of pigments from the fruits. The prickly pear fruit with three different masses (0.5, 1 and 1.5 g) was taken in 250 ml Erlenmeyer flasks. 50 ml of distilled water was used as solvent and it was added to each flask in order to keep the fruit fully immersed in the solvent. Erlenmeyer flask was covered with a plastic wrap during the experiments to prevent evaporation of solvent. The flasks containing fruits along with solvent were incubated at different temperatures (30, 40, 50 °C) and the extracts were taken at different time intervals (20, 70, 120 min). After extraction for a selected time, the mixture was centrifuged at 2268 g for 15 min (Remi R-24 Centrifuge, India) and supernatant liquid was collected for the determination of pigments from the prickly pear fruits. Experiments were performed randomized in order to minimize the effects of unexplained variability in the observed responses due to extraneous factors. All the experiments were performed in triplicate and the average value was used for the determination of pigments (betacyanin and betaxanthin) from the prickly pear.

2.3. Total betacyanin and betaxanthin content

A 10 ml of the extracts was mixed with 10 ml of 50% aqueous methanol solution and the solution was agitated (22.7 g) at room temperature for 30 min (GeNei SLM-IN-OS-16, India). After they were stirred, the samples were centrifuged at 4731 g for 15 min (Remi R-24 Centrifuge, India). Supernatants were filtered through a filter paper (Whatman no. 1, Whatman International Ltd., Maidstone, England), and the extracts obtained were analyzed spectrophotometrically (Shimadzu UV-1800. Kyoto, Spectrophotometric measurements were performed in triplicate, and the betaxanthin and betacyanin content was calculated [8]. Betaxanthin and betacvanin were assessed as indicaxanthin and betanin equivalents and they were computed by the following equation

$$BS(mg/100 g) = \frac{A \times DF \times MW \times 100}{\varepsilon \times 1}$$
 (1)

where BS is betacyanin or betaxanthin, A is the absorption value at the absorption maximum corrected by the absorption at 600 nm, MW is the molecular weight (indicaxanthin = 308 g/mol and betanin = 550 g/mol), ε is the molar extinction coefficient (indicaxanthin = 48,000 L mol⁻¹ cm⁻¹ and betanin = 60,000 L mol⁻¹ cm⁻¹), and 1 is the path length (1 cm) of the cuvette.

2.4. Experimental design

Response surface methodology (RSM) is an empirical statistical modeling technique employed for multiple regression analysis using quantitative data obtained from properly designed experiments to solve multivariate equations simultaneously [32]. RSM is an effective statistical technique for optimizing complex processes. RSM reduces the number of experimental trials required to evaluate multiple parameters and their interactions. In this study RSM was used to optimize and study the effect of independent variables such as extraction temperature, extraction time and mass of the fruits on the extraction of pigments (betacyanin (BC) and betaxanthin (BX)) from the prickly pear fruits. The experiments were established based on a Box-Behnken Design (BBD) with three factors at three levels and each independent variable was coded at three levels between +1, 0 and -1 corresponded to the low level, mid level and high level (Table 1). Coding of the variables was done according to the following equation:

$$x_i = \frac{X_i - X_{cp}}{\Delta X_i}$$
 $i = 1, 2, 3...k$ (2)

where x_i , the dimensionless value of an independent variable; X_i , the real value of an independent variable; X_{cp} , the real value of an

Table 1 Independent variables and their levels used for Box—Behnken design.

Variables, unit	Factors	Levels	Levels		
	X	-1	0	1	
Extraction temperature, Temp (°C)	<i>X</i> ₁	30	40	50	
Extraction time, Time (min)	X_2	20	70	120	
Mass of the fruit, Mass (g)	X_3	0.5	1.0	1.5	

independent variable at the center point; and ΔX_i , step change of the real value of the variable i corresponding to a variation of a unit for the dimensionless value of the variable i.

The number of experiments (N) required for the development of BBD is defined as $N = 2k(k-1) + C_0$ (where k is number of factors and C_0 is the number of central point). The design included 17 experiments and with 5 central points (used to determine the experimental error). The performance of the process was evaluated by analyzing the responses (Y), which depend on the input factors x_1 , x_2 ..., x_k , and the relationship between the response and the input process parameters is described by

$$Y = f(x_1, x_2...x_k) + e$$
 (3)

where f is the real response function the format of which is unknown and e is the error which describes the differentiation.

The behavior of the response surface was investigated for the response function (Y_i) using the second-order polynomial equation. The generalized response surface model is

$$Y = \beta_0 + \sum_{j=1}^k \beta_j x_j + \sum_{j=1}^k \beta_{jj} x_j^2 + \sum_i \sum_{j=2}^k \beta_{ij} x_i x_j + e_i$$
 (4)

where *Y* is the response; x_i and x_j are variables (i and j range from 1 to k); β_0 is the model intercept coefficient; β_j , β_{jj} and β_{ij} are interaction coefficients of linear, quadratic and the second-order terms, respectively; k is the number of independent parameters (k = 3 in this study); and e_i is the error [33].

2.5. Statistical analysis

The statistical analysis was performed using Design Expert Statistical Software package 8.0.7.1 (Stat Ease Inc., Minneapolis, USA). The experimental data were analyzed using multiple regressions and the significance of regression coefficients was evaluated by F-test. Modeling was started with a quadratic model including linear, squared and interaction terms and the model adequacies were checked in terms of the values of R^2 , adjusted R^2 and prediction error sum of squares (PRESS). The significant terms in the model were found by Pareto analysis of variance (ANOVA) for each response and ANOVA tables were generated. The regression coefficients were used to make statistical calculations to generate response surface plots from the regression models.

Table 2Box—Behnken experimental design matrix and experimental responses.

Run Temperature (°C)	Time (min)	Mass (g)	Betacyanin (mg/100 g)			Betaxanthin (mg/100 g)			
			Y_{exp}	Y_{pre}	% Error	Y _{exp}	Y _{pre}	% Error	
1	30	20	1	11.89	11.96	-0.58	18.72	18.83	-0.57
2	30	70	1.5	12.73	12.67	0.44	22.35	22.25	0.44
3	30	120	1	13.04	13.00	0.34	21.77	21.63	0.66
4	50	120	1	13.22	13.15	0.52	22.57	22.47	0.46
5 ^a	40	70	1	12.96	12.96	0.00	22.5	22.50	0.00
6 ^a	40	70	1	12.96	12.96	0.00	22.5	22.50	0.00
7	40	120	1.5	13.29	13.39	-0.75	24.27	24.51	-1.01
8	50	70	1.5	12.76	12.73	0.24	22	21.86	0.62
9	30	70	0.5	11.36	11.39	-0.28	18.85	18.99	-0.73
10 ^a	40	70	1	12.96	12.96	0.00	22.5	22.50	0.00
11	50	20	1	11.86	11.90	-0.37	19.62	19.77	-0.74
12	40	20	1.5	11.54	11.53	0.11	20.72	20.71	0.03
13	50	70	0.5	11.38	11.44	-0.49	21.05	21.15	-0.48
14	40	20	0.5	11.06	10.96	0.90	20.02	19.78	1.21
15 ^a	40	70	1	12.96	12.96	0.00	22.5	22.50	0.00
16	40	120	0.5	11.37	11.38	-0.11	21.47	21.48	-0.03
17 ^a	40	70	1	12.96	12.96	0.00	22.5	22.50	0.00

^a Central points (used to determine the experimental error).

2.6. Verification of optimized conditions and predictive models

Optimal conditions for the extraction of pigments from prickly pear fruit depend on extraction temperature, time and mass of fruit were obtained by Derringer's desired function methodology. The pigments (betacyanin and betaxanthin) content was determined after extraction under the optimal conditions. The experimental and predicted values were compared in order to determine the validity of the models.

3. Results and discussions

3.1. Box-Behnken analysis

The response surface methodology consists of an empirical modeling technique, which has been used to evaluate the relationship between the experimental and the predicted results [34]. To obtain a proper model for the optimization of the extraction process, the BBD, which is generally the best design for response surface optimization, was selected with three process variables (extraction temperature, extraction time and mass of the fruit) at three levels. Experiments were performed according to the experimental design in order to search the optimum conditions and study the effect of process variables on the extraction of pigments (betacyanin and betaxanthin) from prickly pear fruits. The predicted values were obtained by a model fitting technique using the Design Expert software version 8.0.7.1 and proved to be sufficiently correlated with the observed values. The experimental results and predicted values were exhibited in Table 2.

Fitting of the data to various models (linear, interactive, quadratic and cubic models) was carried out to obtain the regression equations. To decide about the adequacy of models among various models to represent the extraction of pigments from prickly pear fruits, two different tests namely sequential model sum of squares and model summary statistics were carried out in the present study [35], and the results are given in Table 3. The adequacy of model summary output (Table 3) indicates that, the quadratic model is statistically highly significant for the present extraction process. The sequential sum squares showed that the *p*-value was lower than 0.0001 for quadratic and cubic models. Model summary statistics showed that, quadratic model was found to have maximum "Adjusted *R*-Squared" and the "Predicted *R*-Squared" values. Cubic model was not recommended for the

Table 3 Adequacy of the model tested.

Source	Sum of squa	res D	F Mean squa	re F value	Prob > F	Remarks	
Sequential model sum of squares for betacyanin							
Mean	2601.53		1 2601.53				
Linear	5.93		3 1.98	6.42	0.0067		
2FI	0.53		3 0.18	0.51	0.6857		
Quadratic	3.43		3 1.14	191.32	< 0.0001	Suggested	
Cubic	0.04		3 0.01	6.4E + 07	< 0.0001	Aliased	
Residual	0		4 0				
Total	2611.4704	1	7 153.62				
Sequential model sum of squares for betaxanthin							
Mean	7875.89		1 7875.89				
Linear	24.60	;	8.20	9.24	0.0015		
2FI	2.73		3 0.91	1.03	0.4191		
Quadratic	8.57		3 2.86	83.09	< 0.0001	Suggested	
Cubic	0.24		3 0.08	6.4E + 07	< 0.0001	Aliased	
Residual	0		4 0				
Total	7912.0307	1	7 465.41				
Source	Std. Dev.	R ²	Adjusted R ²	Predicted R	² PRESS	Remarks	
Model sum	mary statistic	s for be	etacyanin				
Linear	0.56	0.597	0.504	0.325	6.71		
2FI	0.59	0.650	0.440	-0.096	10.89		
Quadratic	0.08	0.996	0.990	0.933	0.67	Suggested	
Cubic	0	1	1	_	+	Aliased	
Model summary statistics for betaxanthin							
Linear	0.94	0.681	0.607	0.459	19.56		
2FI	0.94	0.756	0.610	0.276	26.17		
Quadratic	0.19	0.993	0.985	0.893	3.85	Suggested	
Cubic	0	1	1	_	+	Aliased	

⁺ case(s) with leverage of 1.0000: PRESS statistic not defined.

present extraction process as the Box—Behnken matrix has sufficient data to interpret the outcome of the present system [36]. Therefore, quadratic model was chosen for further analysis.

3.2. Fitting of second order polynomial equations and statistical analysis

The empirical relationship between the experimental results obtained on the basis of Box—Behnken experimental design model and the input variables were expressed by a second-order polynomial equation with interaction terms. The final equations obtained in terms of coded factors were given below

Betacyanin =
$$12.96 + 0.025X_1 + 0.57X_2 + 0.645X_3$$

 $+ 0.052X_1X_2 + 2.5E - 003X_1X_3 + 0.36X_2X_3$
 $- 0.11X_1^2 - 0.35X_2^2 - 0.79X_3^2$ (5)

$$\begin{split} Betaxanthin &= 22.5 + 0.44X_1 + 1.38X_2 + 0.99X_3 + 0.025X_1X_2 \\ &\quad + 0.64X_1X_3 + 0.53X_2X_3 - 1.19X_1^2 - 0.64X_2^2 - 0.24X_3^2 \end{split} \tag{6}$$

However, the exploration and optimization of a fitted response surface might produce poor or misleading results [37]. So, it was necessary to check if the model exhibited a good fit or not. The adequacy and fitness of the models were tested by regression analysis and Pareto analysis of variance (ANOVA). The results indicated that the equation adequately represented the actual relationship between the independent variables and responses (Table 4). ANOVA is a statistical technique that subdivides the total variation in a set of data into component parts associated with specific sources of variation for the purpose of testing hypotheses on the parameters of the model [38]. Analysis of variance followed by Fisher's statistical test (*F*-test) was applied to evaluate the significance of each variable. The *F*-value is the ratio of the mean

Table 4ANOVA analysis and statistical parameters of the model.

Source	Coefficient	Sum of	DF	Standard	Mean	F-value	p-value
	estimate	squares		error	square		•
Betacyanin							
Model	12.96	9.89	9	0.03	1.10	183.77	< 0.0001
X_1	0.03	0.01	1	0.03	0.01	0.84	0.3910
X_2	0.57	2.61	1	0.03	2.61	436.40	< 0.0001
X_3	0.64	3.32	1	0.03	3.32	554.20	< 0.0001
X_{12}	0.05	0.01	1	0.04	0.01	1.84	0.2167
X ₁₃	0.002	0.00	1	0.04	0.00	0.00	0.9503
X_{23}	0.36	0.52	1	0.04	0.52	86.66	< 0.0001
X_1^2	-0.11	0.05	1	0.04	0.05	8.13	0.0246
X_2^2	-0.35	0.52	1	0.04	0.52	86.22	< 0.0001
X_3^2	-0.80	2.66	1	0.04	2.66	444.85	< 0.0001
Residual error		0.04	7		0.01		
Lack of fit		0.04	3		0.01		
Mean	12.37						
C.V. %	0.63						
Adeq. Precision	40.96						
Betaxanthin							
Model	22.50	35.90	9	0.08	3.99	116.04	< 0.0001
X_1	0.44	1.58	1	0.07	1.58	45.83	0.0003
X_2	1.38	15.13	1	0.07	15.13	440.00	< 0.0001
X_3	0.99	7.90	1	0.07	7.90	229.83	< 0.0001
X_{12}	-0.03	0.00	1	0.09	0.00	0.07	0.7952
X ₁₃	-0.64	1.63	1	0.09	1.63	47.29	0.0002
X_{23}	0.53	1.10	1	0.09	1.10	32.07	0.0008
X_{1}^{2}	-1.19	6.00	1	0.09	6.00	174.55	< 0.0001
X_2^2	-0.64	1.70	1	0.09	1.70	49.58	0.0002
X_3^2	-0.24	0.25	1	0.09	0.25	7.28	0.0307
Residual error		0.24	7		0.03		
Lack of Fit		0.24	3		0.08		
Mean	21.52						
C.V. %	0.86						
Adeq. Precision	40.00						

square due to regression to the mean square due to real error and indicates the influence (significance) of each controlled factor on the tested model [34]. The ANOVA results (Table 4) for BC and BX show Fisher F-value of 183.77 and 116.04, which implies that the model is significant and higher. The large value of F indicates that most of the variation in the response can be explained by the regression equation. The associated P value is used to estimate whether P is large enough to indicate statistical significance and P values lower than 0.05 indicate that the model and the terms are statistically significant [39].

Coefficient of determination (R^2) and adjusted- R^2 were also calculated to check the adequacy and fitness of the model. The R^2 gives the proportion of the total variation in the response predicted by the model, indicating the ratio of the sum of squares due to regression (SSR) to the total sum of squares (SST) [40]. The values of R^2 were calculated to be 0.996 and 0.993 for BC and BX respectively, which imply that 95% of experimental data was compatible. A high R^2 coefficient ensures a satisfactory adjustment of the quadratic model to the experimental data. The use of an adjusted- R^2 is to evaluate the model adequacy and fitness. The adjusted- R^2 value corrects the R^2 value for the sample size and for the number of terms in the model. The value of adjusted- R^2 (0.990 for BC and 0.985 for BX) is also high and advocates a high correlation between the observed and the predicted values. The very small p-value (<0.0001) and high coefficient of determination ($R^2 = 0.996$ for BC and $R^2 = 0.993$ for BX) also show that the quadratic polynomial model is significant and sufficient to represent the actual relationship between the response and independent variables.

The coefficient of variation (CV) indicates the relative dispersion of the experimental points from the predictions of the SOP models. As a general rule, the CV should not be greater than 10% and a high CV indicates that variation in the mean value is high and does not

satisfactorily develop an adequate response model [41]. The very low value (0.63 and 0.86) of CV clearly representing a very high degree of precision and a good reliability of conducted experiments. Adequate precision measures the signal to noise ratio and compares the range of the predicted values at the design points to the average prediction error. The ratio greater than 4 is desirable and indicates adequate model discrimination [42]. In this work the ratio is found to be >39, which indicates an adequate signal. Therefore, quadratic model can be used to navigate the design space.

3.3. Adequacy of the models

Generally, it is important to confirm the fitted model to make sure that it gives a sufficient approximation to the actual values. Unless the model shows a satisfactory fit, proceeding with an investigation and optimization of the fitted response surface likely gives poor or misleading results [43]. Diagnostic plots such as the predicted versus experimental values (Fig. 1) help us to judge the model satisfactoriness and exhibit the relationship between predicted and experimental values. In this figure, each of the observed values is compared to the predicted value calculated from the model. The data points on this plot lie reasonably close to the straight line and indicate an adequate agreement between the real data and the data obtained from the models. The result suggests that the models used in this research were able to identify

Y predicted (mg/100g) [®] 13.5 13.0 12.5 12.0 11.5 11.0 10.5 11.5 12.0 12.5 11.0 13.0 13.5 Y experimental (mg/100g) Y predicted (mg/100g) σ 23.6 22.2 20.8 19.4 18.0 19.4 20.8 22.2 23.6 25.0 18.0 Y experimental (mg/100g)

Fig. 1. Comparison between predicted and experimental values of betacyanin (a) and betaxanthin (b).

operating conditions for selective extraction of pigments from prickly pear fruit.

Data were also analyzed to check the normality of the residuals. Normal probability plot represents the normal distribution of the residuals and the residual gives the difference between the observed value of a response measurement and the value that is fitted under the theorized model, and the small residual value indicates that model prediction is accurate [44]. By constructing a normal probability plot of the residuals, a check was made for the normality assumption as shown in Fig. 2 and the data points on this plot lie reasonably close to a straight line. But, some scatter is expected even with normal data, therefore, it could be concluded that the data was normally distributed.

3.4. Effect of process variables on the extraction of pigments

Pigments from the prickly pear fruit were extracted by the aqueous extraction method and the extraction process was carried out with different extraction temperatures, different extraction times and with different mass of the fruits with an explicit objective of determining optimum extraction conditions. 3D response surface and contour plots were used to represent the effect of process variables on the extraction of pigments from prickly pear fruit. The response surface and contour plots showed the relative effects of any two variables when the remaining variable was kept as constant. The response surface plots estimating the specific

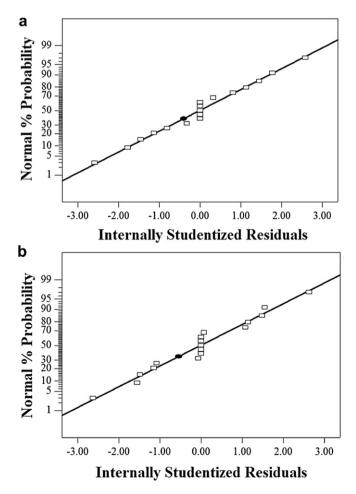


Fig. 2. Normal probability plots of studentized residuals for betacyanin (a) and betaxanthin (b).

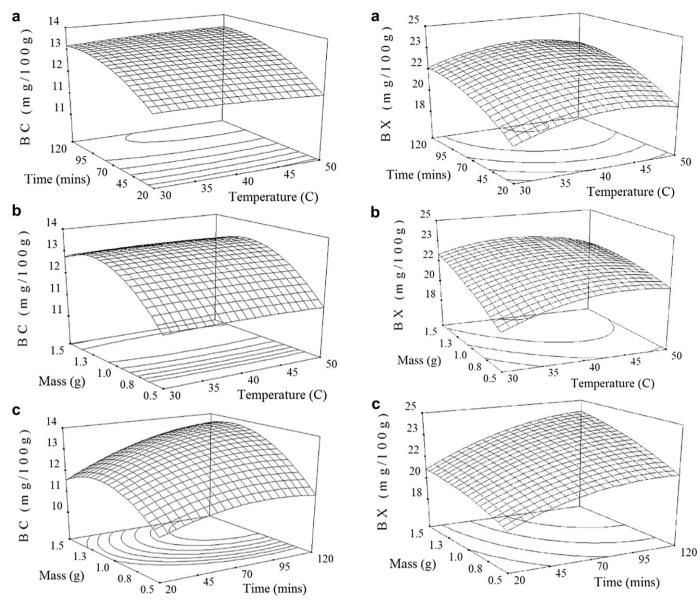


Fig. 3. Response surface plots (3D) showing the effects of variables on the betacyanin yield.

Fig. 4. Response surface plots (3D) showing the effects of variables on the betaxanthin yield.

surface area of pigments extraction versus independent variables are presented in Figs. 3 and 4.

Fig. 3 shows the effects of temperature, time and mass on the extraction of betacyanin pigments from the prickly pear fruit. Extraction time and mass of the fruit were significantly (p < 0.0001) affected the total betacyanin yield in both linear and quadratic manners. Both independent variables showed positive effects in linear terms but showed a negative effect on its quadratic terms. The interactive effects between time and mass showed a positive effects and also significant (p < 0.0001). Pigments extraction from the fruit depends on both the amount of fruit used for extraction (0.5-1.5 g) as well as the extraction time (20-120 min) (Fig. 3c). Fig. 3a shows that the yield of betacyanin increased with the increasing of extraction temperature and extraction time. Woo et al. [45] showed that, the elevated temperature of 50 °C did not show any effect on the stability of betalain pigments from red dragon fruits. The higher temperature would cause softening of the plant tissue, disrupting the

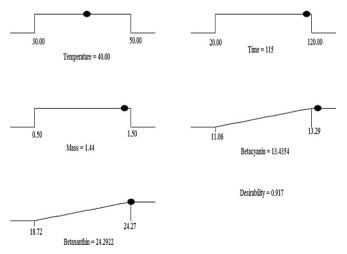


Fig. 5. Desirability ramp for optimization.

Table 5Predicted and experimental values of the responses at optimum conditions

Optimal levels of process parameters	Optimized values ^a (prec	Optimized values ^a (predicted values)		Experimental values ^b		
	BC (mg/100 g)	ng/100 g) BX (mg/100 g) BC (BX (mg/100 g)		
Extraction temperature ($^{\circ}$ C) = 40	13.4354	22.2922	13.32 ± 0.15	24.09 ± 0.21		
Extraction time (min) = 115						
Mass of fruit $(g) = 1.44$						

a Predicted using response surface quadratic model.

interactions between phenolic compounds and protein or polysaccharides and increasing the solubility of the phenolic compounds, which improves the rate of diffusion, thus giving a higher rate of extraction [46]. The response surface plot (Fig. 3b) demonstrates the interaction effect of the process variables (extraction temperature and mass of fruits) on the betacyanin extraction process. Both the variables have a strong effect on the extraction efficiency and the total amount of betacyanin yield was increased with an increase in both the temperature as well as the mass of fruits used for the extraction. It can be seen that (Table 4) the positive linear effect and negative quadratic effects of time and mass explained the observed nature of the curve.

From Table 4, it may be observed that betaxanthin content depends on the extraction time and mass of the fruit. The linear effect of the extraction time and mass were positive (p < 0.0001) whereas its quadratic effect was negative at p < 0.05, which resulted in a curvilinear increase in betaxanthin yield. The interactive effect between the extraction time and mass is positive at p < 0.001. Fig. 4a shows the effect temperature and time on the extraction of betaxanthin from prickly pear. By increasing the temperature and time, total betaxanthin content is increased significantly. Extraction temperature had a close relationship with the extraction of betaxanthin pigments. So with the increase of temperature from 30 to 50 °C, total betaxanthin yield can be increased with the increase of mass of fruits (Fig. 4b). The betaxanthin content is increased with increasing mass and time, which is probably due to the fact that the solvent can enter into the high mass of cells while more pigments can permeate into the solvent under the higher extraction time and mass of the fruits (Fig. 4c).

3.5. Selection of optimum conditions

An optimum condition for the extraction of pigments from prickly pear was determined to obtain maximum betacyanin content and betaxanthin content. Second order polynomial models obtained in this study were utilized for each response in order to obtain specified optimum conditions. The Derringer's desirability function method was employed to optimize the process variables. This function searches for a combination of factor levels that jointly optimize a set of responses by satisfying the requirements for each response in the design. The optimization is accomplished by: converting each response Y_i (i = 1, 2, ..., m) into a dimensionless desirability scale that defines a partial desirability function (d_i) , combining the individual desirabilities to obtain the composite or global desirability function (D), and finally maximizing the D and identifying the optimal factor settings. The scale of the desirability function ranges between 0 (completely undesirable response) and 1 (fully desired response) [47].

The individual desirabilities (*d*) for each response are obtained by specifying the goals, i.e., minimize, maximize or target the response, and boundaries required for each one. A weight factor, which defines the shape of the desirability function for each response, is then assigned. Weights must be between 0.1 and 10, with larger weights corresponding to more important responses.

A weight factor of 1 was chosen for all individual desirabilities in this work. The "importance" of a goal can be changed in relation to the other goals. It can range from 1 (least importance) to 5 (most important). The default is for all goals to be equally important in a setting of 3.

In order to optimize extraction of pigments from prickly pear the following constraints have taken (1) Extraction temperature (30 °C–50 °C), (2) Extraction time (20–120 min) and (3) Mass of fruit (0.5–1.5 g) respectively, were set for maximum desirability. In numerical optimization, the desired goal was preferred for each variable and response from the menu. The possible goals were: within range (for three independent variables) and maximum (for responses only). Applying the methodology of the desired function, the optimum level of various parameters was obtained and it indicates that extraction temperature of 40 °C, Time of 115 min and mass of 1.44 g give 13.4354 mg/100 g of betacyanin content and 24.2922 mg/100 g of betaxanthin content respectively with overall desirability value of 0.917. Fig. 5 showed a ramp desirability that was developed from optimum points via numerical optimization.

3.6. Verification of optimized conditions and predictive model

The suitability of the model equations for predicting optimum response values was tested under the conditions: extraction temperature 40 °C, Time 115 min and mass of 1.44 g. This set of conditions were determined to be optimum by the RSM optimization approach and to confirm the validity of the optimized conditions; experiments were carried out to compare the experimental results with the predicted values of the responses using the model equation. The experiments were conducted in triplicates and the average values were reported in Table 5. The efficiency on the extraction of pigments (betacyanin and betaxanthin) under these optimum conditions was found to be 13.4354 mg/100 g and 24.2922 mg/100 g, and the experimental value $13.32 \pm 0.15 \text{ mg}/100 \text{ g}$ and $24.09 \pm 0.21 \text{ mg}/100 \text{ g}$. The mean values of the pigment content obtained were compared with the predicted values and indicating the suitability of the developed quadratic models. The percentage deviation of the experimental and theoretical results was found as 0.87 and 0.84% respectively. The pigments obtained through confirmation experiments are within 95% of predicted values. This indicates the suitability of the developed quadratic models and it may be noted that these optimal values are valid within the specified range of process parameters.

4. Conclusion

In this study, the statistical methodology, Box—Behnken Response Surface design is demonstrated to be effective and reliable in finding the optimal conditions for the extraction of pigments from prickly pear fruits. The results showed that, the extraction conditions have significant effects on the extraction of pigments. The response surface plots were used for estimating the interactive effect of three independent variables (extraction

 $^{^{\}rm b}$ Mean \pm standard deviation of triplicate determinations from experiments.

temperature, °C; extraction time, min; and mass of fruit, g) on the responses (betacyanin and betaxanthin). Second order polynomial models were developed for predicting betacyanin and betaxanthin content. Analysis of variance showed a high coefficient of determination value (R^2) of 0.996 for betacyanin and 0.993 for betaxanthin ensuring a satisfactory fit of the second-order polynomial regression model with the experimental data. The optimum conditions for extraction of pigments from prickly pear fruit were found to be, extraction temperature of 40 °C, extraction time of 115 min and mass of 1.44 g have been selected to obtain the maximum yield of pigments (betacyanin and betaxanthin) of 13.435 mg/100 g and 22.2922 mg/100 g. Under these optimized conditions the experimental values of pigments agreed closely with the predicted yield.

References

- [1] Russell CE, Felker P. The prickly-pears (*Opuntia* spp. *Cactaceae*): a source of human and animal food in semiarid regions. Econ Bot 1987;41:433—45.
- [2] Pimienta BE. Prickly pear (Opuntia spp.). A valuable fruit crop for the semi-arid lands of Mexico. J Arid Environ 1994;28:1–11.
- [3] Yahia EM, Jacobo CM. Nutritional components and anti-oxidant capacity of ten cultivars and lines of cactus pear fruit (*Opuntia* spp.). Food Res Int 2011;44: 2311–8.
- [4] Castellanos-Santiago E, Yahia EM. Identification and quantification of betalains from fruit of 10 Mexican prickly pear cultivars by high-performance liquid chromatography and electrospray ionization mass spectrometry. J Agric Food Chem 2008;56:5758–64.
- [5] Gandía-Herrero F, Jiménez-Atiénzar M, Cabanes J, García-Carmona F, Escribano J. Stabilization of the bioactive pigment of Opuntia fruits through maltodextrin encapsulation. J Agric Food Chem 2010;58(10):646–52.
- [6] Stintzing FC, Herbach KM, Moßhammer MR, Carle R, Yi WG, Sellapan S. Color betalain pattern and antioxidant properties of cactus pear. J Agric Food Chem 2005;53:442–51.
- [7] Chavez-Santoscoy RA, Gutierrez-Uribe JA, Serna-Saldívar SO. Phenolic composition, antioxidant capacity and in vitro cancer cell cytotoxicity of nine prickly pear (*Opuntia* spp.) juices. Plant Foods Hum Nutr 2009;64(2):146–52.
- [8] Moßhammer MR, Maier C, Stintzing FC, Carle R. Impact of thermal treatment and storage on color of yellow-orange cactus pear (Opuntia ficus-indica [L.] Mill. cv. 'Gialla') juices. J Food Sci 2006;71(7):400-6.
- [9] El-Samahy S, El-Hady Abd, Habiba R, Moussa T. Chemical and rheological characteristics of orange-yellow cactus-pear pulp from Egypt. J Prof Assoc Cactus 2006;8:39–51.
- [10] Felker P, Rodriguez S, Casoliba RM, Filippini R, Medina D, Zapata R. Comparison of Opuntia ficus-indica varieties of Mexican and Argentine origin for fruit yield and quality in Argentina. J Arid Environ 2005;60:405–22.
- [11] Saenz C. Processing technologies: an alternative for cactus pear (*Opuntia* spp.) fruits and cladodes. J Arid Environ 2000;46:209–25.
- [12] Pompeu DR, Silva EM, Rogez H. Optimization of the solvent extraction of phenolic antioxidants from fruits of *Euterpe oleracea* using response surface methodology. Bioresour Technol 2009;100:6076–82.
- [13] Wettasinghe M, Shahidi F. Evening primrose meal: a source of natural antioxidants and scavenger of hydrogen peroxide and oxygen-derived free radicals. J Agric Food Chem 1999;47:1801–12.
- [14] Cacace JE, Mazza G. Optimization of extraction of anthocyanins from black currants with aqueous ethanol. J Food Sci 2003;68:240–8.
- [15] Cacace JE, Mazza G. Extraction of anthocyanins and other phenolics from black currants with sulfured water. J Agric Food Chem 2002;50:5939–46.
- [16] Bujić-Kojić A, Planinic M, Srécko T, Jakabek L, Seruga M. Influence of solvent and temperature on extraction of phenolic compounds from grape seed, antioxidant activity and colour of extract. Int J Food Sci Tech 2009;44: 2394–401.
- [17] Pinelo M, Rubilar M, Sineiro J, Nunez MJ. Effect of solvent, temperature, and solvent-to-solid ration the total phenolic content and antiradical activity of extracts from different components from grape pomace. J Agric Food Chem 2005:53:2111-7.
- [18] Santos DT, Veggi PC, Meireles AMA. Extraction of antioxidant compounds from jabuticaba (*Myrciaria cauliflora*) skins. Yield, composition and economical evaluation. J Food Eng 2010;101:23–31.
- [19] Hsu SY, Chung HY. Effects of j-carrageenan, salt, phosphates and fat on qualities of low fat emulsified meatballs. J Food Eng 2001;47(2):115–21.
- [20] Jakobsen M, Bertelsen G. Colour stability and lipid oxidation of fresh beef. Development of a response surface model for predicting the effects of temperature, storage time, and modified atmosphere composition. Meat Sci 2000;54(1):49–57.
- [21] Jin G, Hea L, Zhang J, Yu X, Wang J, Huang F. Effects of temperature and NaCl percentage on lipid oxidation in pork muscle and exploration of the controlling method using response surface methodology (RSM). Food Chem 2012;131(1):817–25.

- [22] Kumar A, Prasad B, Mishra IM. Process parametric study for ethane carboxylic acid removal onto powder activated carbon using Box—Behnken design. Chem Eng Technol 2007;30(7):932—7.
- [23] Parajo JC, Santos V, Dominguez H, Vazquez M. NH₄OH-based pretreatment for improving the nutritional quality of single-cell protein (SCP). Appl Biochem Biotechnol 1995;55:133—50.
- [24] Ge Y, Ni Y, Yan H, Chen Y, Cai T. Optimization of the supercritical fluid extraction of natural vitamin E from wheat germ using response surface methodology. J Food Sci 2002:67:239–43.
- [25] Hayouni EA, Abedrabba M, Bouix M, Hamdi M. The effects of solvents and extraction method on the phenolic contents and biological activities in vitro of Tunisian Quercus coccifera L. and Juniperus phoenicea L. fruits extracts. Food Chem 2007;105(3):1126–34.
- [26] Huang W, Xue A, Niu H, Zhen J, Wang J. Optimised ultrasonic-assisted extraction of flavonoids from *Folium eucommiae* and evaluation of antioxidant activity in multi-test systems in vitro. Food Chem 2009:114:1147–54.
- [27] Yang Z, Zhai W. Optimization of microwave-assisted extraction of anthocyanins from purple corn (*Zea mays L.*) cob and identification with HPLC-MS. Innov Food Sci Emerg Technol 2010;11:470–6.
- [28] Gao L, Mazza G. Extraction of anthocyanin pigments from purple sunflower hulls. J Food Sci 1996;61:600—3.
- [29] Butera D, Tesoriere L, Gaudio FD, Bongiorno A, Allegra M, Pintaudi AM, et al. Antioxidant activities of Sicilian prickly pear (*Opuntia ficus indica*) fruit extracts and reducing properties of its betalains: betanin and indicaxanthin. I Agric Food Chem 2002:50(23):6895—901.
- [30] Frati AC, Xilotl Diaz N, Altamirano P, Ariza R, Lopez-Ledesma R. The effect of two sequential doses of *Opuntia streptacantha* upon glycemia. Arch Invest Med (Mex) 1991;22(3–4):333–6.
- [31] Tesoriere L, Butera D, Pintaudi AM, Allegra M, Livrea MA. Supplementation with cactus pear (*Opuntia ficus-indica*) fruit decreases oxidative stress in healthy humans: a comparative study with vitamin C. Am J Clin Nutr 2004;80: 391–5.
- [32] Rao JK, Chul-Ho K, Rhee SK. Statistical optimization of medium for the production of recombinant hirudin from Saccharomyces cerevisiae using response surface methodology. Process Biochem 2000;35:639–47.
- [33] Sridhar R, Sivakumar V, Prakash Maran J, Prince Immanuel V. Development of model for treatment of pulp and paper industry bleaching effluent using response surface methodology. Environ Prog Sustain Energy 2011. http://dx.doi.org/10.1002/ep.10581.
- [34] Rožić L, Novaković T, Petrović S. Modeling and optimization process parameters of acid activation of bentonite by response surface methodology. Appl Clay Sci 2010;48:154–8.
- [35] Muthukumar M, Mohan D, Rajendran M. Optimization of mix proportions of mineral aggregates using Box Behnken design of experiments. Cem Concr Compos 2003;25:751–8.
- [36] Kumar A, Prasad B, Mishra IM. Optimization of process parameters for acrylonitrile removal by a low-cost adsorbent using Box-Behnken design. J Hazard Mater 2008;150:174–82.
- [37] Omwamba M, Hu Q. Antioxidant capacity and antioxidative compounds in barley (*Hordeum vulgare L*.) grain optimized using response surface methodology in hot air roasting. Eur Food Res Technol 2009;229:907–14.
- [38] Huiping L, Guoqun Z, Shanting N, Yiguo L. Technologic parameter optimization of gas quenching process using response surface method. Comp Mater Sci 2007;38(3):561–70.
- [39] Segurola J, Allen NS, Edge M, Mahon AM. Design of eutectic photoinitiator blends for UV/visible curable acrylated printing inks and coatings. Prog Org Coat 1999;37:23–37.
- [40] Ghafari S, Aziz HA, Isa MH, Zinatizadeh AA. Application of response surface methodology (RSM) to optimize coagulation—flocculation treatment of leachate using poly-aluminum chloride (PAC) and alum. J Hazard Mater 2009; 162-650. 6
- [41] Koocheki A, Taherian AR, Razavi SMA, Bostan A. Response surface methodology for optimization of extraction yield, viscosity, hue and emulsion stability of mucilage extracted from *Lepidium perfoliatum* seeds. Food Hydrocolloid 2009;23:2369—79.
- [42] Beg Q, Sahai V, Gupta R. Statistical media optimization and alkaline protease production from *Bacillus mojavensis* in a bioreactor. Process Biochem 2003;39: 203–9.
- [43] Murugesan K, Dhamija A, Nam IH, Kim YM, Chang YS. Decolourization of reactive black 5 by laccase: optimization by response surface methodology. Dyes Pigm 2007;75(1):176–84.
- [44] Mohajeri L, Aziz HA, Isa MH, Zahed MA. Statistical experiment design approach for optimizing biodegradation of weathered crude oil in coastal sediments. Bioresour Technol 2010;101:893–900.
- [45] Woo KK, Ngou FH, Ngo LS, Soong WK, Tang PY. Stability of betalain pigment from red dragon fruit (*Hylocereus polyrhizus*). Am J Food Tech 2011;6(2): 140–8.
- [46] Shi J, Yu J, Pohorly J, Young C, Bryan M, Wu Y. Optimization of the extraction of polyphenols from grape seed meal by aqueous ethanol solution. J Food Agric Environ 2003;1:42–7.
- [47] Gonzalez A, Foster KL, Hanrahan G. Method development and validation for optimized separation of benzo[a]pyrene—quinone isomers using liquid chromatography—mass spectrometry and chemometric response surface methodology. J Chromatogr A 2007;1167(2):135–42.